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Combined Gas Law Chart Answer This  
gas law is known as the combined gas  
law, and its mathematical form is.  
$$(P_1 V_1 T_2) = (P_2 V_2 T_1)$$
 This allows us to follow  
changes in all three major Combined Gas  
Law Chart Worksheet

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Answers: COMBINED GAS LAW  
Remember to convert all temperatures  
to Kelvin.  $P_1 V_1 T_1 = P_2 V_2 T_2$   
1.5 atm 3.0 L 20. C 293K 2.5 atm 1.9 L 30. C  
303K 2 720 torr 256 mL 25 C 298 K  
8.0x10<sup>2</sup> torr 250 mL 50. C 323 K 3 600.

mmHg 2.5 L 22 C 295 K 760 mmHg 1.8 L  
270 K 4 1.2 atm 750 mL 0.0 C 273.0 K  
2.0 atm 500. mL 25 C 298 K 5 95 kPa 4.0  
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Combined Gas Law Chart Answer Key

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Combined Gas Law Worksheet - Solutions 1) If I initially have 4.0 L of a gas at a pressure of 1.1 atm, what will the volume be if I increase the pressure to 3.4 atm?  $(1.1 \text{ atm})(4.0 \text{ L}) = (3.4 \text{ atm})(x \text{ L})$   $x = 1.29 \text{ L}$  2) A toy balloon has

an internal pressure of 1.05 atm and a volume of 5.0 L.

### **Combined Gas Law Worksheet**

The combined gas law combines the three gas laws: Boyle's Law, Charles' Law, and Gay-Lussac's Law. It states that the ratio of the product of pressure and volume and the absolute temperature of a gas is equal to a constant. When Avogadro's law is added to the combined gas law, the ideal gas law results. Unlike the named gas laws, the combined gas law doesn't have an official discoverer.

### **Combined Gas Law Definition and Examples**

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### **Answer Key**

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1) A sample of sulfur dioxide occupies a volume of 652 mL at 40° C and 720 mm Hg What volume will the sulfur dioxide occupy at STP? 2) A sample of argon has a volume of 50 dm<sup>3</sup> and the pressure is 0.92 atm If the ...