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answered on a separate sheet of paper
1) Differentiate between matter and
energy. 2) Explain how each of the 4
types of energy are different. 3) Draw an
atom of Carbon and label its Protons,

Neutrons, and Electrons. 4) Name the 4 most abundant elements found in the Human Body. 5) Compare and contrast Ionic and Covalent Bonds.

Chapter 2: Basic Chemistry Name: Period

Figure 2—1 is a diagram of an atom. Select two different colors and use them to color the coding circles and corresponding structures on the figure. Complete this exercise by responding to the questions that follow, referring to the atom in this figure. Insert your answers in the answer blanks provided. O Nucleus Electrons 1. 2. 3. 5. Figure ...

Mr. Burghardt Ponderosa High School Shingle Springs, CA
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Basic Chemistry Concepts of Matter and Energy Composition of Matter Molecules and Compounds Chemical Bonds and Chemical Reactions Biochemistry. Concepts of Matter and Energy. Matter. Matter is anything that occupies space and has weight. Matter exists in three states: gas, liquid, and solid.

Chapter 2: Basic Chemistry

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1 Chapter 2.1, 2.2 Review Packet -
Answer Key 2.1 The Nature of Matter
Lesson Summary Atoms The atom is the
basic unit of matter, made up of three
subatomic particles. Protons have a
positive charge and neutrons carry no
charge. Strong forces bind protons and
neutrons together in the nucleus. An
electron is a negatively charged particle
that has only about 1/1840 the mass of a
proton.