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STOICHIOMETRY

PRACTICE- Review

\u0026 Stoichiometry

Extra Help Problems

Stoichiometry: Limiting

\u0026 Excess

Reactant Unit 6 -

Stoichiometry: Mixed

Problems Mixed

Stoichiometry

Problems Mixed

Stoichiometry p-

Mixed-Stoichiometry

Predicting The Products

of Chemical Reactions -

Chemistry Examples

and Practice Problems

chem 9 3 mixed

stoichiometry problems

Stoichiometry Basic

Introduction, Mole to

Mole, Grams to Grams,

Mole Ratio Practice

Problems **Balancing**

Chemical Equations

Practice Problems

Mixed Stoichiometry

Worksheet Answers

ANSWER KEY. Mixed

Stoichiometry

Problems . 1. $2\text{H}_2 + \text{O}_2$

($2\text{H}_2\text{O}$. a). How many

moles of H_2 would be

required to produce 5.0

moles of water? 5.0
 moles water. b). What
 mass of H_2O is formed
 when H_2 reacts with
 384 g of O_2 ? 432g H_2 .
 2. $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow$
 $\text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$. a).
 Balance this equation.
 Look above. b).

Mixed Stoichiometry Problems

ANSWER KEY Mixed
 Stoichiometry
 Problems 1. $2\text{H}_2 + \text{O}_2 \rightarrow$
 $2\text{H}_2\text{O}$ a). How many
 moles of H_2 would be
 required to produce 5.0
 moles of water? 5.0
 moles water b). What
 mass of H_2O is
 formed when H_2
 reacts with 384 g of O_2 ?
 432g H_2 2. H_2SO_4
 $+ 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4$
 $+ 2\text{H}_2\text{O}$ a). Balance
 this equation Look
 above. b).

Mixed Stoichiometry Problems.docx - Mixed Stoichiometry

...

Mixed Stoichiometry
 Problems. 1. Hydrogen
 and oxygen react
 under a specific set of
 conditions to produce
 water according to. the
 following: $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow$
 $2\text{H}_2\text{O}(\text{g})$ A.

Mixed Stoichiometry Problems -

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Stoichiometry: Mixed
 Problems (KEY) 1) $\text{N}_2 +$
 $3\text{H}_2 \rightarrow 2\text{NH}_3$ What
 volume of NH_3 at STP is
 produced if 25.0 of
 N_2 is reacted with an
 excess of H_2 ? 3 3 3 2 3
 2 2 2 40.0L NH_3 1mol
 NH_3 22.4L NH_3 1mol N_2
 2mol NH_3 28.0g N_2 25.0g
 N_2 1mol $\text{N}_2 \times \frac{2\text{mol NH}_3}{1\text{mol N}_2} \times \frac{17\text{g N}_2}{28\text{g N}_2} =$

Stoichiometry: Mixed Problems (KEY)

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excess of H_2 ? 3 3 3 2 3
2 2 2 40.0L NH_3 1mol
 NH_3 22.4L NH_3 1mol N
2mol

Chemistry If8766 Worksheet Answers

STOICHIOMETRY:
MOLE-MOLE PROBLEMS
I. $\text{N}_2 + 3\text{H}_2$ Name How many moles of hydrogen are needed to completely react with two moles of nitrogen? 2.0 + 302
How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate? (y
owls 3 00 KC/03 3. Zn
+ HCl ZnCl_2 + How many moles of hydrogen are produced from the reaction of three moles

SchoolNotes 2.0

Stoichiometry Worksheet and Key
1.65 mol KClO_3 3 mol
 KClO_3 3 mol O_2 = mol O_2

2 3.50mol KCl = mol
 KClO 3 = 0.275 mol Fe
 = mol Fe 2O 3 = =

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 ($2\text{H}_2\text{O}$. a). How many
 moles of H_2 would be
 required to produce 5.0
 moles of water? 5.0
 moles water. b). What
 mass of H_2O is formed
 when H_2 reacts with
 384 g of O_2 ? 432g H_2 .
 2. $\text{H}_2\text{SO}_4 + 2\text{NaOH}$ (
 $\text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$. a).
 Balance this equation.
 Look above. b).

Mixed Stoichiometry Worksheet Answers

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Mixed Stoichiometry Practice Worksheet Answers ...use stoichiometry to convert moles &/or grams of one reactant &/or product into moles &/or grams of different reactants &/or products...apply stoichiometric conversions to predict which reactant will be the limiting and which will be the excess when given starting amounts of multiple

Mixed Stoichiometry Practice

Stoichiometry - Mixed

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Mole/Mole and
Mole/Mass
Stoichiometry ...**

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Topic: Stoichiometry