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Express your answer as a chemical equation. Identify all of the phases in your answer. $\text{SO}_3(\text{g}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_2\text{SO}_4(\text{aq})$ Part A Predict the products of the following reaction:

$\text{MgCl}_2 + \text{K} \rightleftharpoons$ The equation may not be correctly balanced. Express your answer as a chemical formula. If no reaction occurs, enter noreaction.
 $\text{MgCl}_2 + \text{K} \rightleftharpoons \text{KCl} + \text{Mg}$

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