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Any polyatomic ion

with the suffix “-ate”

uses the suffix “-ic” as

an acid. So, HNO_3 will be nitric acid. When you have a polyatomic ion with one more oxygen than the “-ate” ion, then your acid will have the prefix “per-” and the suffix “-ic.” For example, the chlorate ion is ClO_3^- . Therefore, HClO_4 is called perchloric acid.

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