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CHAPTER 9 Chemical Names And Formulas Test Review ...

Chapter 9 Chemical Names and Formulas83 SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268-270) This section explains the rules for naming and writing formulas for binary molecular compounds. Naming Binary Molecular Compounds (pages 268-269) 1. Circle the letter of the

type(s) of elements that form binary molecular compounds.

Name Date Class CHEMICAL NAMES AND FORMULAS 9

SECTION 9.2 NAMING AND WRITING FORMULAS FOR IONIC COMPOUNDS I. Write the formulas for these binary ionic compounds. c. potassium iodide 2. write the formulas for the compounds formed c. b. f. 3. Name the following binary ionic compounds. sodium sulfide g. CuCl: h. SnCl_2 a. MnO_2 c. d. SrBr_2 221 Chapter 9 Chemical Names and Formulas

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1. Chapter 9 "Chemical Names and Formulas" H2O 1. 2. Section 9.1 Naming Ions OBJECTIVES: • Identify the charges on monatomic ions by using the periodic table, and name the ions. • Define a polyatomic ion and write the names and formulas of the most common polyatomic ions. • Identify the two common endings for the names of most polyatomic ions. Atoms and Ions • Atoms are electrically neutral.

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CHEMISTRY - Chapter 9 - Scotch Plains-Fanwood High School Page 1 Chapters 9 - Chemical Names/Formulas IONS = Charged particles conduct e- in solution = (aq) CATIONS = + ANIONS = - Ionic Compounds - conduct electricity when (aq) or (l) Many [ionic] formula units all attached. ex. NaCl Molecular Compounds - do NOT conduct electricity Individual distinct molecules. ex. H2O

Chapter 9 Packet - Chapters 9 Chemical Names/Formulas

...

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Answer Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253

OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1

DIF: L1 REF: p. 254

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

The name of a base is the name of the cation followed by the name of the anion (hydroxide). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and Formulas of Acids hydrogen1.