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Chapter 12 Stoichiometry Test Answer Key A In any stoichiometry problem, the first step is always to calculate the number of moles of each reactant present. In this case, we are given the mass of K 2 Cr 2 O 7 in 1 mL of solution, which we can use to calculate the number of moles of K 2 Cr 2 O 7 contained in 1 mL:

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