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In chemistry, an oxidizing agent (oxidant, oxidizer), or oxidising agent (oxidiser) is a substance that has the ability to oxidize other substances — in

other words to accept their electrons. Common oxidizing agents are oxygen, hydrogen peroxide and the halogens.. In one sense, an oxidizing agent is a chemical species that undergoes a chemical reaction in which it gains one or more electrons.

**Oxidizing agent - Wikipedia**

The reaction of an organic halide with magnesium is not a Grignard reaction, but provides a Grignard reagent. [5] Grignard reactions and reagents were discovered by and are named after the French chemist François Auguste Victor Grignard ( University of Nancy , France), who published it in 1900 and was awarded the 1912 Nobel Prize in Chemistry for this work.