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 $6.02 \times 10^{23} \text{ atoms/mol} \times 2.62 \times 10^{24} = 1.57 \times 10^{25}$ (b) #
electrons added = $Q/e = 1.00 \times 10^{-3} \text{ C} / 1.60 \times 10^{-19} \text{ C/electron}$
 $= 6.25 \times 10^{15}$ or 2.38 electrons for every 10⁹ already present
23.2 (a) $F_e = k_e q_1 q_2 / r^2 = 8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2 \times 1.60 \times 10^{-19} \text{ C}^2 / (3.80 \times 10^{-10} \text{ m})^2 = 1.59 \times 10^{-9} \text{ N}$ (repulsion) (b) $F_g = G m_1 m_2 / r^2$

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- A particle with charge $Q = 5.00 \text{ C}$ is located at... Ch. 23 - A
particle with charge Q is located at the center... Ch. 23 - (a) A
panicle with charge q is located a distance... Ch. 23 - Find the net
electric flux through (a) the closed...

A thin rod of length l and uniform charge per unit length

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