
Weapon Matrix New World

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substance whose particles are dissolved in a solution. The substance in which the solute dissolves is

called the solvent. For example, seawater is a solution in which salt is the solute and water is the solvent. Solutes and solvents can take the form of a solid, liquid, or gas. The solution takes the state of the solvent. Figure 2 lists some common

Section 8.1 8.1 Formation of Solutions

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Section 8 1 Formation Of Solutions Worksheet Answer Key

Describe how a substance can dissolve in water by dissociation, dispersion, or ionization. Describe how the physical properties of a solution can differ from those of its solute and solvent Identify

energy changes that occur during the formation of a solution Describe factors affecting the rate at which a solute dissolves in a solvent Vocabulary: solute solvent dissociation dispersion ...

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The overall enthalpy change that accompanies the formation of a solution, ΔH_{soln} , is the sum of the enthalpy change for breaking the intermolecular interactions in both the solvent and the solute and the enthalpy change for the formation of new solute-solvent interactions. Exothermic ($\Delta H_{\text{soln}} < 0$) processes favor solution formation. In addition, the change in entropy, the degree of disorder of the system, must be considered when predicting whether a solution will form.