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Separation of a

Mixture Lab **Organic**
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Extractions (part 1)

Experiment 1 Preview
Virtual Chemistry
Experiment: Exploring
Density Part 1

Experiment #1:

Density of Liquids and
Solids - SMU Chemistry

Chemistry 110,
Experiment 8: Video 1 -
Introduction and Part A
CH127—Experiment 7—
Chemical Reactions

Experiment #8:

Heats of Reaction -
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Chem 2115

Experiment 1

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CHEM 2115

Experiment # 1 -

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consistent to the centigram place (i.e. hundredths place) when rounding the milligram balance reading to two decimal places? 2. Volume in graduated cylinders: 3. Density (Attach sample calculations of each type ...

CHEM 2115 LAB REPORT Experiment #1 Mass, Volume and ...

CHEM 2115LAB REPORT Experiment #1Mass, Volume and Density Chem I Lab. Name. Section # Station # Date ... 1. Mass of perforated (Gooch) crucible: Question: Are the two measured masses consistent to the centigram place (i.e. hundredths place) when rounding the milligram balance reading to two decimal

places? Observed
volume. 10 mL
graduated cylinder 50
mL graduated cylinder
2. Volume in ...

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Station # Date This
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observed using
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**CHEM 2115 :
CHEMISTRY I LAB -
Stockton University
1 CHEM 2115 LAB
REPORT Experiment**

#1 Mass, Volume and Density Chem I Lab Name Jordanna Mastroddi Section # Station # A4 Date 01/24/17 1. Mass of object next to the balance: Observed mass 3.414g Posted mass 3.415g 2. Mass of 10 mL graduated cylinder: 3. Volume in graduated cylinders: 4.

**chem lab report exp
1 - 1CHEM 2115 LAB
REPORT
Experiment#1 ...**

Chem 2115 is a co-requisite of Chem 2110. Students will perform experiments in measurement, stoichiometry, solutions, chemical equilibrium, and acids & bases; demonstrating principles learned in Chem 2110. Course Policies The following policies have been

adopted by the Chemistry Program and apply to all sections of Chem I Laboratory. 1.

CHEM 2115: General Chemistry I Laboratory - Spring 2014

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Experiment 1.pdf - CHEM 2115 LAB REPORT Experiment#1 Name

...
II - 1 CHEM 2115 Experiment # 2 The Computer in the Laboratory ... On the first line, write a title

(like "Experiment 1 Data Analysis.") 3. Skip a line. 4. To set up the spreadsheet, put "Trial" in cell A3 and "mass (g)" in cell B3. 5. Enter trial numbers 1 to 6 in cells A4 to A9. 6. Enter your measured mass data for Solution A in cells B4 to B9. At this point, your spreadsheet ...

CHEM 2115

Experiment # 2

1 filled maybe - CHEM 2115 LAB REPORT
Experiment#1 Name
Mass Volume and
Density Section 1 Mass
of object next to the
balance Observed
mass Chem I Lab. 1
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1 CHEM 2115 Lab
 Report Experiment #9
 Ideal Gas Law Chem I
 Lab on # 006 Station #
 A2 Date 11/29/2016 I.
 Data Trial 1 Trial 2 Trial
 3 Mass of Mg used (g)
 0.042 0.044 0.043
 Volume of H₂ gas
 collected (L) 0.04260
 0.04302 0.04453
 Temperature of gas
 (Kelvin) 294.1 294.5
 294.8 Atmospheric
 pressure (inHg) 29.89
 29.89 29.89 Height
 difference (mmH₂O)
 174.5 175.1 170.9
 Vapor pressure of H₂
 O (mmHg) 18.8 19.3
 19.6 II.

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