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To calculate an object's density, divide its mass by its volume. If mass is measured in grams and volume in cubic centimeters, the unit of density is grams per cubic centimeter (g/cm3). Calculate the density of each object, and record the answers in the last column of your data table. Label this column "Density (g/cm3).".

Student Exploration: Density Laboratory
Gizmo Warm-up The Density Laboratory Gizmo allows you to measure a variety of objects, then drop them in water (or other

liquid) to see if they sink or float. 1. An object’s mass is the amount of matter it contains. Mass can be measured with a calibrated scale like the one shown in the Gizmo. Drag the first object onto the Scale. (This is object 1.)

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Vocabulary : calorie, calorimeter, joule, specific heat capacity
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 $\text{slope} = \frac{\text{change in } y}{\text{change in } x} \text{ m} = \frac{y}{x} \text{ m} = \frac{(39.0 - 26.0)\text{g}}{(15.0 - 10.0)\text{mL}} \text{ m} = 2.6 \text{ g/mL}$. Since mass is the y-variable and volume is the x-variable, the slope calculation requires that mass be divided by volume. Thus, the slope of the line represents the density of the glass.

VLab: Density: Answers
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