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Kinetics: Initial Rates  
and Integrated Rate  
Laws

To prepare 0.1 N  
Sodium Thiosulphate  
solution and to  
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1 Chemical reactions

1.1 Rate of reaction  
and dependence on  
temperature We will  
once again look at the  
formation of ammonia  
( $\text{NH}_3$ ) from nitrogen  
and hydrogen (see  
section Chemical  
equilibrium of the  
thermodynamics  
chapter). This reaction  
follows the equation:  $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$  (1)  $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$   $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$  To find the Gibbs  
free energy of ...

### **Introduction to Chemical Engineering: Chemical Reaction**

...

Conversion and its  
related terms yield and  
selectivity are  
important terms in

chemical reaction  
engineering. They are  
described as ratios of  
how much of a  
reactant has reacted (X  
— conversion, normally  
between zero and  
one), how much of a  
desired product was  
formed (Y — yield,  
normally also between  
zero and one) and how  
much desired product  
was formed in ratio to  
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product(s) ...

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discussion of PDE as a  
building block for  
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education, research  
and practice seems still  
lacking, preventing a  
broader impact on  
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realities more and  
more concerned with  
the 'knowledge  
economy'.

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engineering — past,  
present, future ...**

After completing

Chapter 5 of the text  
and associated website  
material, the reader  
will be able to:.

Describe the CRE  
algorithm that allows  
the reader to solve  
chemical reaction  
engineering problems  
through logic rather  
than memorization.

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Chemical Reaction  
Engineering**

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overview |  
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A. Tavares da Silva,  
The penetration theory for gas absorption accompanied by a first-order chemical reaction with a diffusing catalyst, Chemical Engineering Science,

10.1016/0009-2509(74)85053-0, 29, 1, (275-279), (1974).

**Gas absorption  
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