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4A1: The rate of a reaction is influenced by the concentration or pressure of reactants,

the phase of the reactants and products, and environmental factors such as temperature and solvent. 4A2: The rate law shows how the rate depends on reactant concentrations.

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reaction. Most people encounter kinetics every day, whether they realize it or not. Take for example, when you open your refrigerator. Refrigerators cool your food to ...

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Factors Affecting the Rate of Chemical Reactions

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There are factors that affect reaction rates including concentration, temperature, surface area, and catalysts.

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Rate of Reaction of Sodium Thiosulfate and Hydrochloric ...

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Temperature affects the rate of chemical reaction, the warmer the water is, the faster the reaction will be; the colder the water is, the slower the reaction would be. Conclusion - Lab 6-b: Factors Affecting Reaction Rate Lab Report Experiment 6 Rates of Chemical Reactions By Nikhola Mirashirova Lab Partner: Dina

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- in my experiment, the time Alka Seltzer tablet uses to dissolve in water decreases as the water becomes hotter and increase as the temperature becomes lower. Temperature affects the rate of

chemical reaction; the higher temperature the reactants have, the higher the rate of chemical reaction will be; the lower

temperature the reactants have, the lower the rate of chemical reaction will be.