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Chemquest 11 An Electron39s Address

11. In your own words, what is the difference between a sublevel and an orbital? Orbitals are contained within sublevels. A sublevel is a grouping of orbitals. 12. How many electrons can fit in each of the following energy levels: 1st energy level = 2 because only 2 can fit in an s sublevel and the first energy level only has an s sublevel.

Information: Energy Levels and Sublevels

Information: Representing the Most Probable Location of an Electron The following is an “address” for an electron—a sort of shorthand notation.

The diagram below represents an electron located in an orbital inside of the p sublevel in the 3rd energy level.
EXAMPLE #1:

Information: Energy Levels and Sublevels

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electrons can fit in each of the following energy levels: 1st energy level = 2 because only 2 can fit in an s sublevel ...

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