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atomic weight of an element is 26.89. If its equivalent weight is 8.9, the exact atomic weight of
element would be (a) 26.89 (b) 8.9 (c) 17.8 (d) 26.7. Answer. Answer: (d) 26.7 Explanation: Atomic
weight = (Equivalent weight × Valency) =(8.9 × 3) = 26.7

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following phenomenon is considered responsible for Global ...

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the surface of the boundary separating two bulk phases. Adsorption is the process of attracting and
holding a substance's molecules on the surface of a liquid or a solid resulting in a higher surface
concentration of the molecules.