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The Nature Of The Hydrogen

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Abstract. Hydrogen bond (H-bond) effects are well known: it makes sea water liquid, joins cellulose microfibrils in sequoia trees, shapes DNA into chromosomes, and polypeptide chains into wool, hair, muscles, or enzymes. However, its very nature is much less known and we may still wonder why O-H...O energies range from less than 1 to more than 30 kcal/mol without evident reason.

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Hydrogen bond (H-bond) effects are known: it makes sea water liquid, joins cellulose microfibrils in trees, shapes DNA into genes and polypeptide chains into wool, hair, muscles or enzymes. Its true nature is less known and we may still wonder why O-H...O bond energies range from less than 1 to more than 30 kcal/mol without apparent reason.

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Hydrogen ion, strictly, the nucleus of a hydrogen atom separated from its accompanying electron. The hydrogen nucleus is made up of a particle carrying a unit positive electric charge, called a proton. The isolated hydrogen ion, represented by the symbol H^+ , is therefore customarily used to represent

hydrogen ion | Definition, Charge, Formula, & Facts ...

It is light, storable, energy-dense, and produces no direct emissions of pollutants or greenhouse gases. But for hydrogen to make a significant contribution to clean energy transitions, it needs to be adopted in sectors where it is almost completely absent, such as

transport, buildings and power generation.

The Future of Hydrogen - Analysis - IEA

A hydrogen bond (often informally abbreviated H-bond) is a primarily electrostatic force of attraction between a hydrogen (H) atom which is covalently bound to a more electronegative atom or group, particularly the second-row elements nitrogen (N), oxygen (O), or fluorine (F)—the hydrogen bond donor (Dn)—and another electronegative atom bearing a lone pair of electrons—the hydrogen bond acceptor (Ac).

Hydrogen bond - Wikipedia

Subsequent to the recent re-definition of hydrogen bonding by the IUPAC committee, there has been a growing search for finding the presence of this ever interesting non-covalent interaction between a hydrogen atom in an X-H group and any other atom in the periodic table. In recent gas phase experiments, it h 1st International Conference on Noncovalent Interactions PCCP Emerging Investigator ...

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Facts About Hydrogen | Live Science

The H₂ uptake ability of H₂-treated soil disappeared when the soil was autoclaved or when glucose was added,

suggesting the biotic nature of the H₂ uptake agent. Physical disturbance reduced the H₂ uptake ability of the

soil, indicating the special colonial or long filamentous structure of the H₂ uptake agent in soil.