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acids hydrochloric acid

and sulfuric acid were

performed the heat

given off by these two

reactions was used to

determine the

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reactants in each experiment

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different amounts of ingredients. Do the same with chemical reactions. See how many products you can make with different amounts of reactants. Play a game to test your understanding of reactants, products and leftovers. Can you get a perfect score on each level?

Reactants, Products and Leftovers - PhET

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Example 1:
Calculate the mass of a magnesium, Mg, atoms in grams. 24.035 g Mg . 1 mol Mg . 1 molecule Mg = 4.04×10^{-23} g/Mg atom
1 mol Mg 6.02×10^{23} molecules
1 atom Mg
Example 2:

Calculate the number of atoms in one-millionth of a gram of magnesium, Mg.

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