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Specific Heat Capacity Questions - Miss Wise's Physics Site

specific heat capacity questions and answers 3 Give that the specific heat capacity of iron as 450JKg⁻¹ K⁻¹, and the heat capacity is 1800JK⁻¹, calculate the mass of the iron.

Physics: specific heat capacity questions and answers ...

The specific heat capacity of water is 4,200 Joules per kilogram per degree Celsius (J/kg°C). This means that it takes 4,200 J to raise the temperature of 1 kg of water by 1°C. Some other examples...

Specific heat capacity - Energy and heating - AQA - GCSE ...

The specific heat capacity of liquid water is 4.184 J/ g degree C. Calculate the quantity of energy in Joule required to heat 2.00 g of water from 11.6 degree C to 87.3 degree C. View Answer. The...

Heat Capacity Questions and Answers | Study.com

TAP 607-4: Specific heat capacity: some questions. TAP 607-5: Thermal changes. TAP 607- 1: Measuring the specific heat capacity of a metal. The value of the specific thermal capacity of a material tells us how much energy is needed to change the temperature of one kilogram of the material by 1 degree.

Specific heat capacity questions and equation

Specific Heat Capacity. This is the change in energy stored in an object or system as its temperature changes. Quick Questions. 2. Specific Heat Capacity: Example A. Here you need to work out the energy required to raise a certain mass of water by a known temperature. 3.

Specific Heat Capacity | GCSE Physics Online

Latent heat and Specific heat capacity questions. 1. How much water at 50°C is needed to just melt 2.2 kg of ice at 0°C? 2. How much water at 32°C is needed to just melt 1.5 kg of ice at -10°C? 3. How much steam at 100° is needed to just melt 5 kg of ice at -15°C? 4. A copper cup holds some cold water at 4°C.

Latent heat and Specific heat capacity questions.

Students use the specific heat capacity of water to work out the energy transferred in some everyday contexts. Thinking deeper. The Southern Hemisphere has more area covered by ocean than the Northern Hemisphere. Suggest what affect this has on the climate of the Southern Hemisphere. Why does sweating cool us down? The specific heat capacity of ethanol is 2.46 J/g o C. Would you rather sweat water or ethanol? Explain your answer.

