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$\text{Al(s)} + \text{OH}^- (\text{aq}) \rightarrow$
 $\text{Al(OH)}_4^- (\text{aq}) +$
 $\text{H}_2(\text{g})$ In this reaction,
Al (s) is oxidized to Al^{3+} , and H^+ in water is
reduced to H_2 gas,
which bubbles through
the solution, agitating
it and breaking up the
clogs. The overall
redox reaction is
composed of a
reduction half-reaction
and an oxidation half-
reaction.

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Reaction and Revolution. In 1848 liberals and nationalists rebelled against many of the conservative governments of Europe. As you read, use a chart like the one below to summarize the causes of the revolutions in France in 1830 and 1848.

Chapter 19, Section 2 Reaction and Revolution

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Section 2. The secretary of health and human services shall appoint, with the approval of the governor, a commissioner of mental health who shall serve at the pleasure of the secretary and may be removed by the secretary at any time, subject to the approval of the governor.

General Law - Part I, Title II, Chapter 19, Section 2

Chapter 19 – Reaction Rates and Equilibrium
Section 19.1 Rates of Reaction 1.1: Explain the collision theory of reactions (q. 38). 19.1 Chemical reactions require collisions with sufficient energy to break and form bonds. 1.2: How is the activation energy of a reaction like a wall or barrier (q. 39)? 19.1

Chapter 19 Reaction Rates and Equilibrium

19.4 Calculation of Entropy Changes. In Section 5.5 we discussed how calorimetry can be used to measure H for chemical reactions. No comparable, easy method exists for measuring S for a reaction. By using experimental measurements of the variation of heat

capacity with temperature, however, we can determine the absolute entropy, S , for many substances at any temperature.

Chemistry: The Central Science, Chapter 19, Section 4

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