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Mastering Chemistry Chapter 4 Answer

See Section 4.1 (Page 103) . ANSWER: Correct Element Names, Symbols, and Atomic Numbers Learning Goal: To identify elements by their chemical symbol and their atomic number. Chemical elements are substances that are composed of only one kind of atom and cannot be broken down into simpler substances.

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soluble salts, the molarity of a particular
ion is equal to the molarity of that
compound times the subscript for that
ion. For example, 1 M of AlCl_3 is 1 M in
 Al^{3+} and 3 M in Cl^- . 1 M of
 $(\text{NH}_4)_2\text{SO}_4$ is 2 M in NH_4^+ and 1 M in
 SO_4^{2-} . Chapter 4 Mastering Chemistry

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