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questions 4 and 5. Check your answers with the Gizmo. (Because dissociation is not always complete, your answers may vary slightly from values in the Gizmo.)  $[\text{H}_2\text{SO}_4] = 0.108 \text{ M}$   $\text{pH H}_2\text{SO}_4 = 0.967$   $[\text{HBr}] = 0.210 \text{ M}$   $\text{pH HBr} = 0.678$  7. Apply: For a strong base, the concentration of hydroxide ions  $[\text{OH}^-]$  is roughly estimated to

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