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Experiments on Properties of Acids-Bases - CBSE Class 10 ...

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Mineral Acids: Mineral acids are acids prepared from minerals. For example, Hydrochloric acid (HCl), Sulphuric Acid (H_2SO_4), and nitric acid (HNO_3) etc. Also Check ⇒ Dilute Acids. Bases. The most common characteristic of bases is their bitter taste and

soapy feel. A base is a substance that renders hydroxyl ion (OH^-) in their aqueous ...

Acids, Bases, and Salts - Introduction, Dissociation ...

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Importance of Acids and Bases | Healthfully

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An acid is a compound that yields hydrogen ion (H^+) in aqueous solution and a base is a compound that gives hydroxyl (OH^-) ions in aqueous solution. As defined by Arrhenius, acid-base reactions are characterized by acids, which dissociate in aqueous solution to form hydrogen ions (H^+) and bases, which form hydroxide (OH^-) ions.