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Pick an element that is not equal on both

sides of the equation. Add a coefficient

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The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry. 2. Stoichiometry is based on the law of conservation of mass. 3. In any chemical reaction, the

mass of the products is less than the mass of the reactants. 4.

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1) What are some signs a chemical reaction has taken place? 2) Give an example of a chemical equation and label the reactants and product. 3) What happens to the atom arrangement when a substance goes through a chemical reaction? 4) Determine how H₂O can put out certain fires, but hydrogen is flammable.

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Reactions/Properties of Matter ...

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Chemistry Chapter 11 Chemical Reactions 11.6: Rates of Reactions Chemical kinetics is the study of the rates of chemical reactions. In this lesson, you will learn how to express the rate of a chemical reaction

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