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$P = \rho g h \Rightarrow$ where ρ (mercury)- $1.36 \times 10^4 \text{ kg/m}^3$, g - 9.81 N/kg , h - 0.76 m . Then $P = (1.36 \times 10^4) \times 9.81 \times 0.76 = 1.014 \times 10^5 \text{ Pa}$. NOTE- this is the standard atmospheric pressure, sometimes called one atmosphere. It is approximately one bar. District mocks, kcse past papers, notes, form 1- 4 papers available on www.kusoma.co.ke.

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